

Synthesis and Characterization of Hydroxyapatite from Duck Eggshell by Wet Precipitation Process



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Keywords: Duck eggshell, Hydroxyapatite, Wet precipitation process, Synthesis, Characterization

ABSTRACT: Hydroxyapatite (HAp) is a natural mineral found in duck eggshells. This study aims to synthesize HAp from duck eggshells using a wet precipitation process. The eggshells were first cleaned and then crushed into a fine powder. The powder was then reacted with a solution of calcium chloride and phosphate ions. The resulting HAp was characterized using X-ray diffraction (XRD) and scanning electron microscopy (SEM). The results show that the synthesized HAp has a similar structure to the natural HAp found in duck eggshells.

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1. INTRODUCTION

Hydroxyapatite (HAp) is a natural mineral found in duck eggshells. This study aims to synthesize HAp from duck eggshells using a wet precipitation process. The eggshells were first cleaned and then crushed into a fine powder. The powder was then reacted with a solution of calcium chloride and phosphate ions. The resulting HAp was characterized using X-ray diffraction (XRD) and scanning electron microscopy (SEM). The results show that the synthesized HAp has a similar structure to the natural HAp found in duck eggshells.

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2. MATERIALS AND METHODS

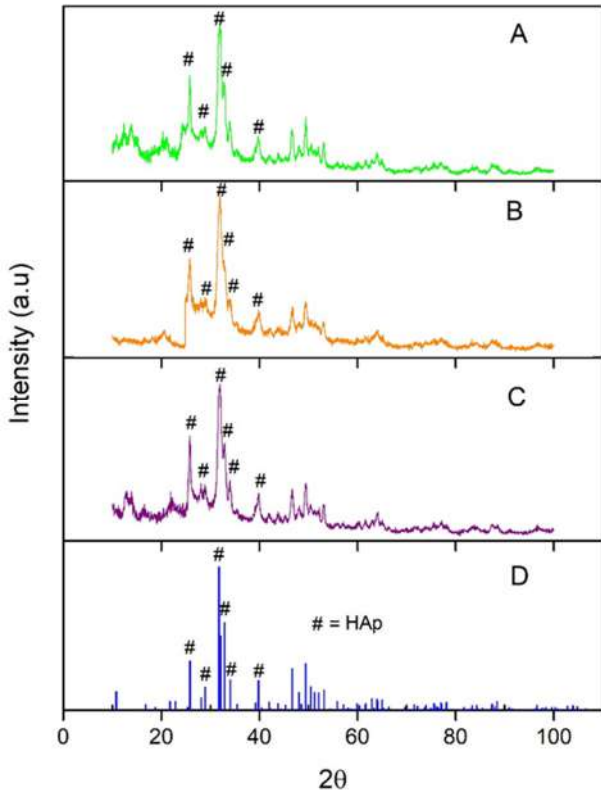
2.1. Materials

The materials used in this study include duck eggshells, calcium chloride, and phosphate ions. The duck eggshells were first cleaned and then crushed into a fine powder. The powder was then reacted with a solution of calcium chloride and phosphate ions. The resulting HAp was characterized using X-ray diffraction (XRD) and scanning electron microscopy (SEM). The results show that the synthesized HAp has a similar structure to the natural HAp found in duck eggshells.

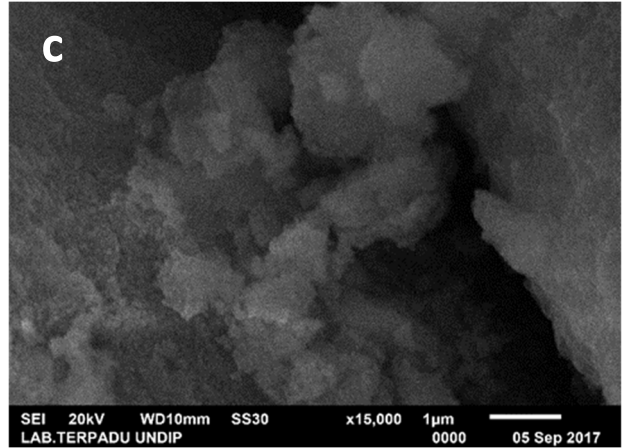
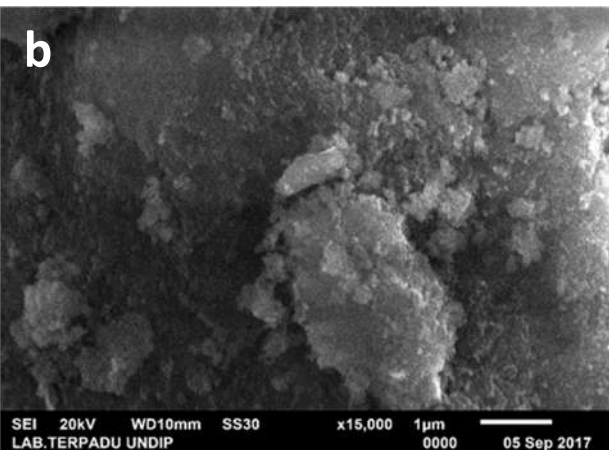
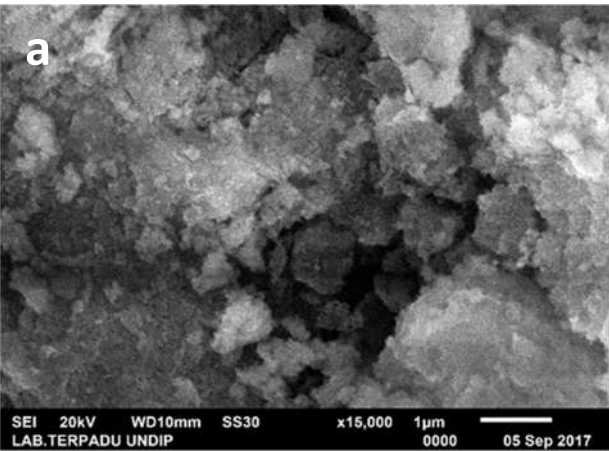


Figure 1: Duck eggshells used in the study.

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Ns 272 dZL ám qz -r P I á á-fi pqóm Rkúzs RÁqp m7440 n- 6940- 644 áy Op- P I á Rkú pnp QL L 454; 8185; 6



Ns 2 82 [MU y wó-sónát -r t fipó-fflná mltth á-fi pqó nlt644 áy nz p Thq y -xno ónth -r Kmnp X óqno Th Thn S2 ; n- S2 ; nz p o- S2@ 2

q puny qThó -r t fipó-fflná mltth oftrRkú fi nRA moffmThp fRAs Thq [ot qóqóRA qé ffnthz 2 q ffnfz -n Thz qp mltthq ffnúthz -r Kmnp X y -xno ónth r-ó644 áy Rkúzs RÁqp0 uRA-fi z mltth nmq S2

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] nmq 6 Kn0X nz p y wó-qxy qz Th-z Thz Thz p Kmnp X ónth u P I á áó-pfóTh

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I x	42@	

q ónth y -xno -r Km np X t fipó-fflná mltth fi nRA Rkú ThqRkú ró-y XKK -r pffow qss RkúRkú nR52 AA2 q Rkúmq náqm -r t fipó-fflná mltth á-fi pqó fi nRA ot nóno Thóttq fRAs J M] nz mRkúRkú p Thq ffnfz -n Thz qp fi nR992A6Ay 63s2

4. CONCLUSION

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